Code No: B1105/R10 m R10

## I B.Pharmacy I Semester Supplementary Examinations, Feb. 2015 PHARM. INORGANIC CHEMISTRY-I

Time: 3 hours Max Marks: 75

## Answer any FIVE Questions All Questions carry equal marks

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- 1. (a) Explain the qualitative tests for the following cations:
  - (i) Zinc (ii) Magnesium (iii) Ferrous
  - (b) Explain the qualitative tests for the following anions:
  - (i) Carbonate (ii) Iodide (iii) Tartarate

[8+7]

- 2. Explain the principle and procedure involved in the limit test for
  - (a) Iron
  - (b) Heavy metals

[8+7]

- 3. (a) What are major physiological ions? Explain the biological importance of sodium and potassium.
  - (b) Write the method of preparation and uses of sodium chloride and potassium chloride. [8+7]
- 4. write short notes on the following
  - (a) Dialysis fluids
  - (b) O R S
  - (c) Biological importance of magnesium

[3x5=15]

- 5. Explain the principle and procedure involved in the assay of
  - (a) Magnesium oxide
  - (b) Sodium acid phosphate
  - (c) Calcium carbonate

[3x5=15]

- 6. Discuss preparation, assay and uses of sodium citrate and  $AL(OH)_3$ . gel [7+8]
- 7. Write short notes on
  - (a) titanium dioxide and ferric oxide.
  - (b) Aluminium stearate
  - (b) bentonite
  - (d) calcium carbonate
- 8. Discuss in detail about
  - (a) talc.
  - (b) bentonite
  - (c) colloidal silica



[4+4+4+3]

[5+5+5]

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